

## INSTITUTE OF NON-FERROUS METALS

## Analytical Chemistry Department 44-101 Gliwice, ul. Sowińskiego 5 CERTIFICATE OF ANALYSIS

Chemical reference material of Galmei ore RG 8

Component	Mean content	Mean deviation, s
Zn	5,40	0,073
Pb	0,84	0,038
Fe	6,34	0,077
Cd	0,047	0,0016
S	0,57	0,018
SiO <sub>2</sub>	2,64	0,10
CaO	26,45	0,14
MgO	12,16	0,12
Al <sub>2</sub> O <sub>3</sub>	0,90	0,039
Zn oxidized	(4,36)	(0,06)
Pb oxidized	(0,72)	(0,02)

The average results of chemical analyses in wt %

(Value s has been calculated for n=20-24)

Director of the Institute Zfrazy Prof. Ph.D. Zbigniew Śmieszek

Analytical methods applied:

Complexometric method in the presence of xylene orange
atomic absorption method
Urbasch method (titration with hexacyangoen potassium)

ferrate (II)

- Pb atomic absorption method
  - polarographic method
- Fe titration manganometric or chromatometric - complexometric in the presence of sulfosalicyclic acid
- d atomic absorption method
- S weight method in form of BaS04, after solving the sample in nitric acid with addition of bromine, or after wasting the Sample with a mixture of sodium potassium carbonate wit h zinc oxide
- Si weight method in form of SiOz

Ca - manganometric titration

- complexometric in the presence of fluorexone with fluoresceine
- Mg weight method in form of Mg\_P\_07
  - complexometric, in the presence of eriochrome black. T
- Al atomic absorption method
  - complexometric, in the presence of xylene orange
  - spectrophotometric method
- Zn oxidized atomic absorption method

Pb oxidized - atomic absorption method

The chemical analyses have been carried out in two industrial laboratories and at the Institute of Non-Ferrous Metals.